# Photochromic Heterocyclic Fulgides. Part 6.<sup>1</sup> Base-catalysed Cyclizations of 1-(2,5-Dimethyl-3-furyl and 2,5-Dimethyl-3-thienyl)ethylidene(isopropylidene)-succinic Anhydrides and Related Compounds

Sean L. Bowden, Stuart A. Harris, Harry G. Heller\* and Michael J. E. Hewlins School of Chemistry and Applied Chemistry, University of Wales College of Cardiff, PO Box 912, Cardiff CF1 3TB, Wales

The title fulgides **1a** and **3a** (X = O or S) rearrange in boiling pyridine to yield deep red *cis*-8,8adihydro-2,4,7,8a-tetramethyl-7*H*-cyclohepta[*b*]-furan and -thiophene-5,6-dicarboxylic anhydrides **6a** (X = O or S) respectively. The corresponding 2-thienyl fulgide **8** undergoes a similar rearrangement but with a subsequent 1,5-shift of its 3-hydrogen to give the colourless *cis*-5,6dihydro-5,8-dimethyl-4*H*-cyclohepta[*b*]thiophene-6,7-dicarboxylic anhydride **11**. The 3-thienylfulgide **1d** (X = S) undergoes a similar rearrangement to yield the colourless *trans*-7,8-dihydro-4,7-dimethyl-6*H*-cyclohepta[*b*]thiophene-5,6-dicarboxylic anhydride **7**. 3-Methyl-5,5-diphenylcyclohepta-1(7),2-diene-1,2-dicarboxylic anhydride **13** is obtained in an analogous reaction from (*E*)-3,3-diphenylallylidene(isopropylidene)succinic anhydride **12**. Structural assignments are based on NMR studies. The mechanistic implications of the reactions are discussed.

Photochemical ring-closures of 3-furyl-, 3-thienyl- and pyrrol-3-ylethylidene(isopropylidene)succinic anhydrides (fulgides) **1a** (X = O, S, or NMe) to give highly coloured 7,7a-dihydrobenzofuran, 7,7a-dihydrobenzothiophene and 7,7a-dihydroindole derivatives **2a** (X = O, S and NMe) have been reported earlier.<sup>1</sup>

We report now a new and general base-catalysed rearrangement of furyl and thienyl fulgides to yield dihydrocyclohepta[b]-furan and -thiophenedicarboxylic anhydride [DHCF and DHCT] derivatives respectively. This reaction was discovered accidentally<sup>4</sup> during an attempt to isomerise the Zfulgide **3a**  $(X = O)^2$  in boiling pyridine. The pyridine solution turned deep red and the colour was not discharged on acidification. A single deep red crystalline compound, *cis*-8,8adihydro-2,4,7,8a-tetramethyl-7*H*-cyclohepta[*b*]furan-5,6-di-

carboxylic anhydride **6a** (X = O) was obtained in high yield. Its structure follows from its 360 MHz <sup>1</sup>H NMR spectrum which showed three singlet signals ( $\delta$  2.12, 2.10 and 1.30) assigned to the 2,4 and 8a methyl groups. The 3 H doublet at  $\delta$  1.41 (J 7.5 Hz) (which was assigned to the pseudo equatorial 7-methyl group) coupled to a multiplet at  $\delta$  3.13. The latter was assigned to the pseudo-axial 7-H because of its large coupling (12 Hz) to the pseudo axial 8-H which gave a resonance at  $\delta$ 1.99 and its 8 Hz coupling to the pseudo equatorial 8-H at  $\delta$ 2.37 (See Fig. 1).

It can be seen from molecular models that a pseudo axial-axial arrangement of the 7-methyl and 8a-phenyl substituents in DHCF **6c** (X = O) should result in shielding of the 7-methyl by the 8a-phenyl group. The fulgide **1c** (X = O) was converted into the red 8,8a-DHCF **6c** (X = O) when boiled in pyridine. Its NMR spectrum did not show shielding of the 7-methyl by the 8a-phenyl group, which is in accord with the structural assignment of a pseudo-equatorial methyl in both DHCFs **6a** and **6c** (X = O). The NMR spectrum of the latter showed shielding of the pseudo-axial 7-H and the pseudo equatorial 8-H by the 8a-phenyl group.

(Z)-2,5-Dimethyl-3-thienylfulgides **3a** and **3b**  $(X = S)^3$  rearranged to the red *cis*-8,8a-DHCTS **6a** and **6b**; (X = S) when boiled in pyridine. Detailed NMR studies indicated that the two DHCTs **6a** and **6b** (X = S) have structures analogous to the 8,8a-DHCFs **6a** (X = O) and a *cis*-configuration of their 7 and **8a** hydrogens. The absorption spectra of the red compounds **6a** (X = O) and **6a** (X = S) and the 7,7a-DHBF **2a** (X = O) are shown in Fig. 2.

The pale yellow (E)-1,2,5-trimethylpyrrol-3-ylfulgide **1b**  $(X = NMe)^1$  gave a deep blue colour in boiling pyridine which was discharged on acidification. NMR analysis of the product indicated that the main reaction was E-Z isomerisation to give a 1:1 mixture of E- and Z-fulgides **1b** and **3b** (X = NMe). No other products could be identified.

The (E)-3-thienylfulgide 1d  $(X = S)^3$  rearranged to give the colourless *trans*-7,8-dihydro-4,7-dimethyl-6*H*-cyclohepta-[*b*]thiophene-5,6-dicarboxylic anhydride 7, on boiling in pyridine. Its structure follows from its NMR spectrum which showed a doublet at  $\delta$  2.64 (assigned to the allylic 4-Me) coupled (*J* 2) to a multiplet at  $\delta$  2.81 (pseudo-axial 6-H) which, in turn, was coupled (*J* 10) to a multiplet at  $\delta$  2.95 (assigned to the pseudo-axial 7-H). The large coupling is attributed to an axial-axial arrrangement of 6- and 7-hydrogens. The lowfield doublet at  $\delta$  1.37 (*J* 7) is assigned to the pseudoequatorial 7-Me which lies in the deshielding zone of a carbonyl of the anhydride ring. The assignment of configuration of the 7-Me as pseudo equatorial is supported by the NOE enhancements seen at the protons assigned as 6 $\beta$ -H, 7 $\alpha$ -H and 8 $\beta$ -H on irradiation of the 7-Me.

The (E)-2-thienylfulgide  $\mathbf{8}^3$  showed a similar rearrangement on boiling in pyridine to give the colourless *cis*-5,6-dihydro-5,8-dimethyl-4*H*-cyclohepta[*b*]thiophene-6,7-dicarboxylic

anhydride 11. Its NMR spectrum showed a doublet (J 2) at  $\delta$ 2.80 (assigned to the allylic 8-Me) coupled to the multiplet at  $\delta$  3.65 (assigned to the pseudo-axial 6-H). The latter was coupled (J 3) to the multiplet at  $\delta$  3.02 (assigned to the pseudo-equatorial 5-H). On the basis of these assignments, it follows that the 5-methyl must be pseudo-axial and remote from the deshielding zone of a carbonyl of the anhydride ring. This is reflected in its highfield chemical signal ( $\delta$  0.95) compared to the equatorial 7-methyl ( $\delta$  1.37) in isomer 7. The coupling constant (J 4) between the pseudo-equatorial 5-H and the pseudo-axial 4-H is consistent with the assigned stereochemistry. The structure and stereochemistry was confirmed by X-ray crystallographic analyses by M. B. Hursthouse and A. Karaulov<sup>6</sup> (See Fig. 3).

3-Methyl-5,5-diphenylcyclohepta-1(7),2-diene-1,2-dicarboxylic anhydride 13 was obtained in high yield in an analogous reaction when (E)-3,3-diphenylallylidene(isopropylidene)succinic anhydride 12 was boiled in pyridine.

(Z)-Isopropylidene(phenylethylidene)succinic anhydride 14 did not undergo the above rearrangement on boiling in pyridine



but isomerised to give a 6:3:1 mixture of the (*E*)-fulgide 15 and the two 1,5-H-shift products 16 and 17 respectively. A similar product composition was obtained when the (*Z*)-fulgide 14 was heated in CDCl<sub>3</sub> in a sealed NMR tube at 140 °C.<sup>5</sup>

The first step in the mechanism for the base-catalysed rearrangements described above is considered to be abstraction of a proton from a methyl of the isopropylidene group, followed by an intramolecular 1,6-addition to give a resonance-stabilised enolate anion which can give rise to different products depending on which carbon is reprotonated.

A probable mechanism for the formation of dihydro derivatives 6a-c and 11, is direct protonation of anions 5a-c or 9 from the sterically less hindered side. An alternative mechanism



Fig. 1 Part of 360 MHz <sup>1</sup>H NMR spectrum of *cis*-8,8a-dihydro-2,4,7,8a-tetramethyl-7*H*-cyclohepta[*b*]furan-5,6-dicarboxylic anhydride **6a** (X = O).



**Fig. 2** Absorption spectra of  $ca. 1 \times 10^{-4}$  mol dm<sup>-3</sup> solutions of compounds **2a** (X = O) (A), **6a** (X = O) (B) and **6a** (X = S) (C) in toluene

is protonation to give *cis* dihydro compounds (*e.g.* 10) followed by 1,5-H shift in the case of **6a-c** or two 1,5-H shifts in the case of **6d** and 11).

Compound 7 (X = S), having the *trans*-axial-axial arrangement of 6- and 7a-hydrogens could not be formed by the latter mechanism unless the strained *trans*-4,10-dihydro intermediate was formed initially, which is considered unlikely. The simpler alternative mechanism is that the enolate anion 5 undergoes protonation from the sterically less-hindered side, before or after a 1,5-hydrogen shift has occurred.

#### Experimental

UV spectra were measured in toluene on a Pye Unicam SP8-200 spectrophotometer; NMR spectra were obtained with a Bruker WM360 spectrometer; J values are recorded in Hz. Nuclear Overhauser enhancements were measured at 360 MHz on a Bruker WM360 spectrometer by the difference method using standard Bruker software. Low-intensity presaturation pulses of 5 s duration were applied before each scan. A sequence of 8 pulses at the selected position followed by 8 pulses at a nearby 'blank' position was repeated 24 or 48 times. The summed 'irradiated' and 'blank' free induction decay patterns were subtracted and transformed after processing with a linebroadening factor of 1 Hz. Quantitative values for enhancement were obtained from integrals of the difference spectra. In all cases, 32 K data points were used over a spectral width of 4505 Hz. Mass spectra were recorded by electron impact on a Varian CH5D mass spectrometer.

Procedure for the Base-catalysed Rearrangements of Fulgides.—A 10% w/v solution of the fulgide in anhydrous pyridine



Fig. 3 X-ray structure of compound 11

was refluxed (3 h), cooled and poured into 3 mol  $dm^{-3}$  hydrochloric acid. The aqueous solution was extracted with ether, washed with water, and dried (MgSO<sub>4</sub>) and removed. The residue was crystallised from chloroform-petroleum.

## cis-8,8a-Dihydro-2,4,7,8a-tetramethyl-7H-cyclohepta[b]-

furan-5,6-dicarboxylic Anhydride **6a**, (X = O).—The cis-8,8a-DHCF **6a** (X = O) was obtained in 88% yield from (*E*)- or (*Z*)-1-(2,5-dimethyl-3-furyl)ethylidene(isopropylidene)succinic anhydride<sup>2</sup> **1a** or **3a** (X = O) (1 g), as red plates, m.p. 129– 130 °C. (Found: C, 68.95; H, 6.1%; M<sup>+</sup>, 260. C<sub>15</sub>H<sub>16</sub>O<sub>4</sub> requires C, 69.21; H, 6.19%; *M*, 260);  $v_{max}/cm^{-1}$  1817m, 1741s and



1546s;  $\lambda_{max}/nm$  461 ( $\epsilon/dm^3$  mol<sup>-1</sup> cm<sup>-</sup> 11820). NMR spectrum, see Fig. 2.

cis-8,8a-Dihydro-2,4,7,8a-tetramethyl-7H-cyclohepta[b]thiophene-5,6-dicarboxylic Anhydride **6a** (X = S).—The cis-8,8a-DHCT **6a** (X = S) was obtained in 67% yield from (Z)-1-(2,5dimethyl-3-thienyl)ethylidene(isopropylidene)succinic anhydride<sup>3</sup> **3a** (X = S) (1 g), as red plates, m.p. 142 °C (Found: C, 65.15; H, 5.85%; M<sup>+</sup> 276. C<sub>15</sub>H<sub>16</sub>SO<sub>3</sub> requires C, 65.24; H, 5.79%; M, 276);  $v_{max}/cm^{-1}$  1838m, 1813m and 1757s;  $\lambda_{max}/nm$ 486 (ε/dm<sup>3</sup> mol<sup>-1</sup> cm<sup>-1</sup> 11 670);  $\delta$  1.39 (3 H, d, J 7, 7β-Me), 1.54 (3 H, s, 8aβ-Me), 2.20 (6 H, s, 2-Me, 4-Me), 2.33 (1 H, dd, J 14, 12, 8β-H), 2.40 (1 H, dd, J 14, 8, 8α-H), 3.16 (1 H, ddq, J 12, 8, 7,7α-H) and 6.02 (1 H, s, 3-H).

cis-8,8a-Dihydro-2,7,8a-trimethyl-7H-cyclohepta[b]thiophene-5,6-dicarboxylic Anhydride **6b** (X = S).—The cis-8,8a-DHCT **6b** (X = S) was obtained in 28% yield from (E)-2,5dimethyl-3-thienylmethylene(isopropylidene)succinic anhydride<sup>3</sup> **1b** (X = S) (0.5 g), as red needles, m.p. 138–140 °C (Found: C, 63.85; H, 5.4.  $C_{14}H_{14}SO_3$  requires C, 64.10; H, 5.38%);  $v_{max}/cm^{-1}$  1839m, 1810m and 1753s;  $\lambda_{max}/nm$  476 ( $\varepsilon/dm^3$ mol<sup>-1</sup> cm<sup>-1</sup> 14 640);  $\delta$  6.20 (1 H, s, olefinic), 5.88 (1 H, s, olefinic), 3.15 (1 H, m, methine 7-H), 2.30 (2 H, m, 2 × methylene 8-H), 2.20 (3 H, s, 2-Me), 1.60 (3 H, s, 8a-Me) and 1.44 (3 H, d, J 7, 7-Me).

cis-5,6-Dihydro-5,8-dimethyl-4H-cyclohepta[b]thiophene-6,7dicarboxylic Anhydride 11.—The cis-5,6-DHBT 9 was obtained in 98% yield from (E)-1-(2- thienyl)ethylidene(isopropylidene)succinic anhydride  $8^3$  (0.5 g), as pale yellow plates, m.p. 140–141 °C (Found: C, 62.5; H, 4.98%; M<sup>+</sup>, 248. C<sub>13</sub>H<sub>12</sub>SO<sub>3</sub> requires C, 62.90; H, 4.87%; M, 248); v<sub>max</sub>/cm<sup>-1</sup> 1818s, 1746s and 1609s;  $\lambda_{max}/333$  nm ( $\epsilon/dm^3$  mol<sup>-1</sup> cm<sup>-1</sup> 16 070);  $\delta$  0.95 (3 H, d, J 7, 5 $\alpha$ -CH<sub>3</sub>), 2.80 (3 H, d, J 2, 8-CH<sub>3</sub>), 2.86 (1 H, ddd, J 17, 4, 1, 4 $\alpha$ -H), 3.02 (1 H, qddd, J 7, 6, 4, 3, 5 $\beta$ -H), 3.34 (1 H, dd, J 17, 6, 4 $\beta$ -H), 3.65 (1 H, br, 6 $\beta$ -H), 7.00 (1 H, d, J 5, 3-H) and 7.51 (1 H, d, J 5, 2-H); NOE 5 $\alpha$ -Me $\rightarrow$ 5 $\beta$ -H (10%), 5 $\alpha$ Me $\rightarrow$ 4 $\beta$ -H (5%), 3-H $\rightarrow$ 4 $\alpha$ -H (2.6%) and 3-H $\rightarrow$ 4 $\beta$ -H (1.4%).

trans-7,8-Dihydro-4,7-dimethyl-6H-cyclohepta[b]thiophene-5,6-dicarboxylic Anhydride 7.—The trans-6,7-DHCT 7 was obtained in 93% yield from (E)-1-(3-thienyl)ethylidene-(isopropylidene)succinic anhydride 1d;  $X = S^3$  (0.5 g), as



colourless plates, m.p. 149 °C (Found: C, 62.5; H, 4.95%; M<sup>+</sup>, 248.  $C_{13}H_{12}SO_3$  requires C, 62.90; H, 4.87%; *M*, 248);  $v_{max}/cm^{-1}$  1817m, 1756s and 1618m;  $\lambda_{max}/nm$  312 ( $\varepsilon/dm^3$  mol<sup>-1</sup> cm<sup>-1</sup> 9800);  $\delta$  1.37 (3 H, d, *J* 7, 7\beta-CH<sub>3</sub>), 2.64 (3 H, d, *J* 2, 4-CH<sub>3</sub>), 2.66 (1 H, d, *J ca.* 13, 8β-H), 2.81 (1 H, dq, *J* 10, 2, 6β-H), 2.9–3.0 (2 H complex 8α-H and 7α-H), 7.16 (1 H, d, *J* 6, 3-H) and 7.22 (1 H, d, *J* 5, 2-H); NOE 3-H→4-Me (5%), 7β-Me→8β-H (2.5%), 7β-Me→6β-H (9%) and 7β-Me→7α-H (8%).

cis-8,8a-Dihydro-4,7-dimethyl-2,8a-diphenyl-7H-cyclohepta-[b] furan-5,6-dicarboxylic Anhydride **6c** (X = O).—From (Z)-1-(2,5-diphenyl-3-furyl)ethylidene(isopropylidene)succinic anhydride **3c** (X = O) (0.25 g), cis-8,8a-DHCF **6c** (X = O) was obtained as red plates, m.p. 167–168 °C (Found: C, 78.1; H, 5.35.  $C_{25}H_{20}O_4$  requires C, 78.11, H, 5.24%);  $\delta$  7.67 (2 H, m, aromatics), 7.28–7.28 (8 H, m, aromatics), 6.44 (1 H, s, olefinic 3-H), 3.15 (1 H, dd, J 12 and 7.5 methylene 8-H), 2.84 (1 H, ddq, J 12, 7.5 and 7, methine 7-H), 2.42 (3 H, s, 4-Me), 2.35 (1 H, dd, J 12 and 12, methylene 8-H), and 1.37 (3 H, d, J 7, 7-Me).

5,5-Diphenyl-3-methylcyclohepta-1(7),2-diene-1,2-dicarboxylic Anhydride 13.—The title compound 13 was obtained in 74% yield from (E)-3,3-diphenylallylidene(isopropylidene)succinic anhydride 12 (0.27 g), as colourless needles (from chloroform-light petroleum), m.p. 222 °C (Found: C, 79.75; H,

\* We thank Dr. P. J. Darcy for preparing this fulgide.

5.51%; M<sup>+</sup>, 330.  $C_{22}H_{18}O_3$  requires C, 79.98; H, 5.49%; *M*, 330);  $v_{max}/cm^{-1}$  1812s, 1757s, 1663m and 1645m;  $\delta$  7.2–7.3 (6 H, complex *m*- and *p*-ArH); 7.00 (4 H, d, *J* 7, *o*-ArH), 3.48 (2 H, s, methylene 7-Hs), (2 H, d, *J* 2, methylene 5-Hs) and 2.44 (3 H, s, 1-Me).

Attempted Cyclisation of (E)-1,2,5-Trimethylpyrrol-3-ylmethylene(isopropylidene)succinic Anhydride **1b** (X = NMe).— The fulgide (0.35 g)<sup>1</sup> in pyridine (100 cm<sup>3</sup>) was boiled (4 h). Work-up gave a 1:1 mixture of the (E) and (Z)-fulgides **1b** (X = NMe) and **3b** (X = NMe), no cyclization having occurred.

## (Z)-1-(2,5-Diphenyl-3-furyl)ethylidene(isopropylidene)-

succinic Anhydride 6c (X = O)\*.—A mixture of diethyl isopropylidenesuccinate (5 g) and 3-acetyl-2,5-diphenylfuran (5 g) was added dropwise with stirring to a solution of potassium *tert*-butoxide [prepared by dissolving potassium (3 g) in boiling *tert*-butyl alcohol (100 cm<sup>3</sup>)] at room temperature. After 3 h, most of the solvent was removed under reduced pressure and water added. On addition of 5 mol dm<sup>-3</sup> hydrochloric acid, the half ester separated and was hydrolysed with 10% ethanolic KOH. Solvent was removed and the residue acidified with dilute hydrochloric acid. On treatment of the resulting diacid with acetyl chloride the (Z)-fulgide 6c (X = O) separated and was recrystallised from ethanol to give yellow plates (0.3 g), m.p. 156 °C. (Found: C; 78.1, H, 5.3. C<sub>25</sub>H<sub>20</sub>O<sub>4</sub> requires C, 78.11; H, 5.24%);  $\lambda_{max}/mm 310$  ( $\varepsilon/dm^3 mol^{-1} cm^{-1} 2820$ );  $\delta 2.01$ , 2.21 and 2.50 (9 H three s, three Me s), 6.71 (1 H, s, 4-H) and 7.45-7.90 (10 H, m, ArH).

#### References

- 1 H. G. Heller, S. A. Harris and S. N. Oliver, Part 5, J. Chem. Soc., Perkin Trans. 1, 1991, 3259.
- 2 P. J. Darcy, H. G. Heller, P. J. Strydom and J. Whittall, J. Chem. Soc., Perkin Trans. 1, 1981, 202.
- 3 A. P. Glaze, S. A. Harris, H. G. Heller, W. Johncock, S. N. Oliver, P. J. Strydom and J. Whittall, J. Chem. Soc., Perkin Trans. 1, 1985, 957.
- 4 A. Gabbutt, Ph.D. Thesis 1985, U.C.W., Aberystwyth.
- 5 H. G. Heller, S. N. Oliver and M. Shawe, J. Chem. Soc., Perkin Trans. 1, 1979, 154.
- 6 Personal communication. M. B. Hursthouse and A. Karaulov.

Paper 1/04275K Received 15th August 1991 Accepted 6th December 1991